

Characterization of Novel Cannabinoid Based T-Type Calcium Channel Blockers with Analgesic Effects

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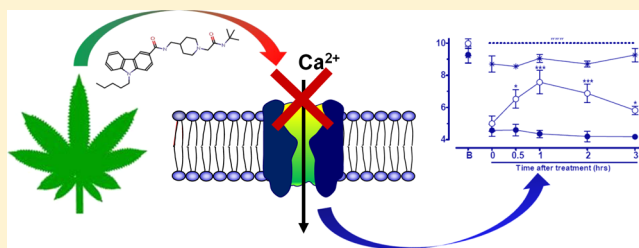
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S Supporting Information

ABSTRACT: Low-voltage-activated (T-type) calcium channels are important regulators of the transmission of nociceptive information in the primary afferent pathway and finding ligands that modulate these channels is a key focus of the drug discovery field. Recently, we characterized a set of novel compounds with mixed cannabinoid receptor/T-type channel blocking activity and examined their analgesic effects in animal models of pain. Here, we have built on these previous findings and synthesized a new series of small organic compounds. We then screened them using whole-cell voltage clamp techniques to identify the most potent T-type calcium channel inhibitors. The two most potent blockers (compounds **9** and **10**) were then characterized using radioligand binding assays to determine their affinity for CB1 and CB2 receptors. The structure–activity relationship and optimization studies have led to the discovery of a new T-type calcium channel blocker, compound **9**. Compound **9** was efficacious in mediating analgesia in mouse models of acute inflammatory pain and in reducing tactile allodynia in the partial nerve ligation model. This compound was shown to be ineffective in Cav3.2 T-type calcium channel null mice at therapeutically relevant concentrations, and it caused no significant motor deficits in open field tests. Taken together, our data reveal a novel class of compounds whose physiological and therapeutic actions are mediated through block of Cav3.2 calcium channels.

KEYWORDS: *hCav 3.2, T-type calcium channel, inflammatory pain, neuropathic pain, carbazole scaffold, electrophysiology*



T-type calcium channels are known for regulating neuronal and cardiac pacemaker activity.^{1–4} They open in response to small membrane depolarizations that in turn trigger the initiation of action potentials.^{1,5,6} Disruption of this sensitive signaling mechanism often leads to hyperexcitability disorders such as arrhythmia, epilepsy, and pain.^{3,7–21} The mammalian genome expresses three different types of T-type calcium channels, Cav3.1, Cav3.2, and Cav3.3, with specific cellular functions.²² The Cav3.2 T-type channel isoform is particularly interesting due to its role in afferent pain signaling. Indeed, up-regulation of the Cav3.2 T-type channel isoform in primary afferent fibers has been linked to chronic pain disorders, whereas ablation of these channels mediates analgesia.^{13,14,17} The development of selective T-type channel antagonists has not been a trivial undertaking, with only a few such small organic molecules having recently been identified.^{15,23–33} Even more challenging is the development of Cav3 isoform selective blockers due to the large degree of sequence similarity among these three Cav3 channel family members.

Many of the known organic molecules that have been shown to modulate T-type calcium channels have structures similar to endogenous anandamide-related molecules named lipoamino acids.^{16,23,34} Lipoamino acids are known to interact with T-type calcium channels and several are also closely related to

endocannabinoids.^{16,25,34,35} Therefore, it is not surprising that many of these T-type blockers also interact with cannabinoid (CB) receptors.^{16,25,34,35} We have previously shown that this mixed T-type/cannabinoid block has beneficial effects in inducing analgesia in animal models of inflammatory pain.^{16,34} However, interactions with CB receptors, particularly CB₁ receptors, can have side effects that may affect mood and memory, in addition to their known psychoactive effects.^{36,37} Synthetic T-type calcium channel antagonists TTAP-1 based on substituted piperidines have been previously disclosed (Scheme 1).

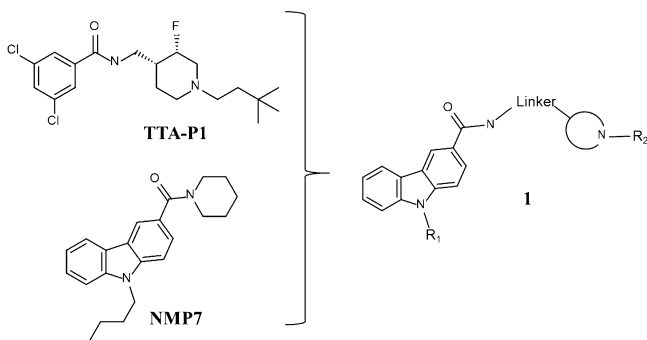
In previous studies, we synthesized and characterized a series of novel cannabinoid ligands with the primary structure bearing a carbazole scaffold such as compound NMP7.^{16,34} These compounds produced mixed cannabinoid receptor/T-type channel blockers that were found to be efficacious in animal models of inflammatory and neuropathic pain. Interestingly, from structure–activity relationships (SARs), we determined that tertiary amines were important for Cav3.2 block^{16,23,34} and

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Scheme 1. Piperidine Containing T-type Ca^{2+} Inhibitors TTA-P1, Dual T-type Channel Blocker/Cannabinoid Agonist NMP7, and Chemical Optimization Plan 1 To Decrease Cannabinoid Receptor Affinities



that the length of the linker attaching the tertiary amine to the carbazole scaffold affected binding to CB_1 and CB_2 receptors.³⁸ Some of these compounds appeared to preferentially and potently inhibit T-type channels *in vitro*. The goal of this study was to optimize Cav3.2 calcium channel selectivity of our series of carbazole derivatives.

Here we used the aforementioned compounds as a starting point for the development of a new series of compounds **1** (Figure 1) based on a carbazole scaffold with an added

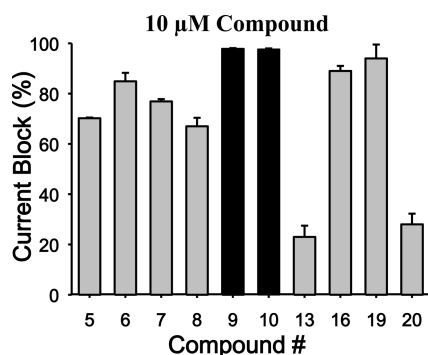


Figure 1. Percentage of whole cell current inhibition of human Cav3.2 (T-type) in response to $10\ \mu\text{M}$ application of the compound series ($n = 6$ per compound). Note the potent and preferential block of Cav3.2 channels by compounds **9** and **10**. Error bars reflect standard errors. For Cav3.2 channels, the holding and test potentials were respectively -110 and -20 mV.

heterocyclic bearing a tertiary amine. We modified the chain length attached to the nitrogen of the carbazole, the length of the linker between the amide bond and the heterocycle ring and introduction of a lipophilic moiety attached to the heterocycle and characterized this set of compounds *in vitro* for their ability to blocking transiently expressed human Cav3.2 (hCav3.2) calcium channels and tested their affinities for cannabinoid receptors. The most potent and selective compound (**9**) was then tested in mouse models of inflammatory and neuropathic pain, revealing potent analgesia by virtue of its Cav3.2 channel blocking ability.

CHEMISTRY

The synthesis of the carbazoles derivatives is outlined in Scheme 2. Amidation under standard peptide coupling conditions³⁸ of *N*-alkylated carbazole-3-carboxylic derivatives

2 with Boc-protected amines afforded the desired amide derivatives **3** and **6**. Deprotection of the Boc protecting group in the presence of TFA in dichloromethane followed by alkylation of the resultant compounds **4** and **8** with *N*-*tert*-butyl-2-chloroacetamide provided the corresponding desired compounds **9**, **10**, **13**, **16**, and **19** (Tables 1 and 2). Compound **20** was prepared by reductive amination of compound **4** using 3,3-dimethylbutyraldehyde (Table 2).

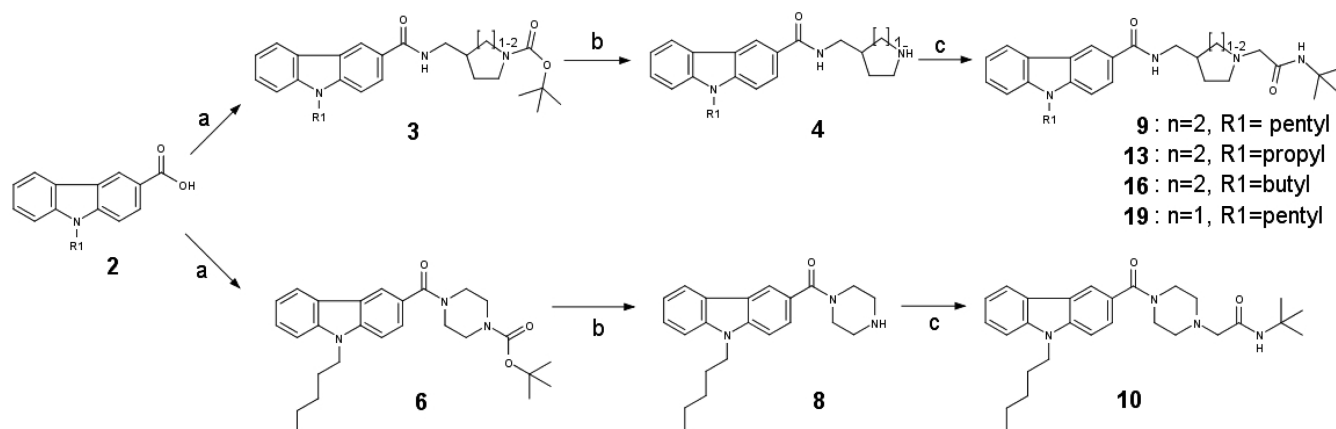
RESULTS

In Vitro Characterization of the Compound Series.

The entire first set of 10 compounds was screened using whole-cell voltage clamp techniques for their ability to mediate tonic block of transiently expressed hCav3.2 channels (Figure 1 and Table 3). Next, we used radioligand binding assays to assess the affinities of these compounds for both CB_1 and CB_2 receptors (Table 1).

In the course of our initial exploratory work on the structure–activity relationship for this novel series of T-type channel blockers, we decided to determine the optimal linker length attached to the carbazole's carbonyl (Table 1). We observed various degrees of inhibition of these channels, with compounds **9** and **10** being the most potent blockers of expressed hCav3.2. These two compounds mediated near complete inhibition at our standard test concentration of $10\ \mu\text{M}$ (Figure 1). Interestingly, both of these compounds are very similar in structure, with both having a cyclic tertiary amine attached to the carbazole scaffold (Table 1). Previous work has indicated that this modification helps confer T-type channel blocking activity onto various organic molecules,^{16,28,29,34} in agreement with our data presented here. As shown in Table 1, compound **10** showed high affinity binding to CB_1 receptors ($15\ \text{nM}$), whereas its affinity for the CB_2 receptor was approximately 100-fold lower ($2\ \mu\text{M}$). Compound **9**, however, did not bind to the two receptors with an affinity less than $5\ \mu\text{M}$ for CB_1 and CB_2 . The only difference between compounds **9** and **10** is the elongated chain attached to the nitrogen of the carbazole in **9** (Table 1). This type of modification has been shown to alter cannabinoid receptor binding,^{16,34,38} but it has not been demonstrated whether this modification affects the interactions of these compounds with Cav3.2. Among the first series of compounds **5–10**, replacement of a piperazine moiety by a methylpiperidine moiety appeared to be the most optimal for decreasing cannabinoid receptor affinity without impacting Cav3.2 calcium channel block. This striking difference in affinity for cannabinoid receptors between compounds bearing a piperazine moiety compared to a methylpiperidine moiety underscores the importance of chain length when developing compounds that preferentially target Cav3.2 calcium channels over cannabinoid receptors. Next, we determined the optimal chain length attached to the carbazole's endocyclic nitrogen (Table 1 and Table 2, compounds **9**, **13**, and **16**). Among the linear *N*-1 alkyl chains, a pentyl chain seemed to be the most optimal for occupying its interaction site Cav3.2 channels, because systematically decreasing the length from *n*-pentyl in **9** negatively impacted the respective Cav3.2 blocking activities. Replacement of the piperidine ring by a pyrrolidine moiety¹⁶ had a slight negative effect on Cav3.2 block, probably due to the lack of optimal ligand–receptor van der Waals contacts. Replacing the amide chain on the piperidine ring by an alkyl chain³⁴ dramatically decreased the Cav3.2 block.

As is clearly seen in the traces in Figure 2A, the slight structural modification in **9** compared to **10** does indeed impact

Scheme 2^a

^aReagents and conditions: (a) corresponding amine, EDAC, HOBt, DMAP, DIPEA, DCM, 0°C to rt; (b) TFA, CH₂Cl₂; (c) *N*-*tert*-butyl-2-chloroacetamide, K₂CO₃, KI, *n*-butanol, reflux 3 h.

Table 1. Radioligand Competitive Binding Assays (mean \pm SEM) for Carbazole-Based Analogues 7–10: Systematic Variation in the Linker Length^a

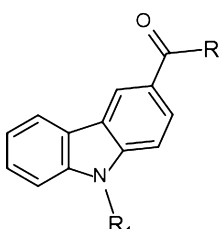
No.	R1	R2	rCB1 K_i^b (nM)	hCB2 K_i^b (nM)
5			n.b.	n.b.
6			>5,000	2,957 \pm 1,362
7			n.b.	n.b.
8			283 \pm 65	2,833 \pm 1305
9			>5,000	>5,000
10			15.0 \pm 6.9	1,968 \pm 906

^aValues are means of three experiments run in triplicate with standard deviation; n.b. no binding.

hCav3.2 channel inhibition. The affinity of **9** versus **10** increased more than 2-fold with the IC₅₀ of **9** and **10** being 1.48 and 3.68 μ M, respectively (Figure 2B and Table 3). In addition, compound **10** shifted the half activation potential of

hCav3.2 by -12 mV (Figure 2C and Table 3). There was no significant effect on half-inactivation potential ($P = 0.143$) (Figure 2D and Table 3). We then tested the Cav3 channel subtype selectivity of compound **9** using a single concentration

Table 2. Analogues of Compound 9: Systematic Variation in N-Alkyl Chain Length and in the Region Occupied by the Heterocycle



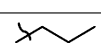
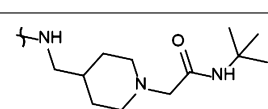
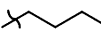
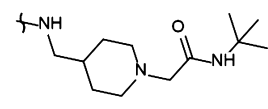
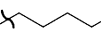
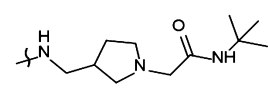
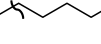
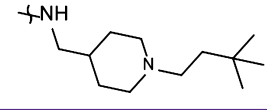
No.	R1	R2
13		
16		
19		
20		

Table 3. Summary of Biophysical Parameters of hCav3.2 Calcium Channel in the Absence and the Presence of Compounds 10 and 9^a

	$V_{0.5act}$ (mV) 3 μ M	V_h (mV) 3 μ M	IC_{50} tonic (μ M)
Wt hCav3.2	-30.0	-53.1 \pm 1.67	
compound 9	-29.7	-58.2 \pm 1.43	1.48 \pm 0.2
compound 10	-42.0*	-58.1 \pm 1.31	3.68 \pm 0.5

^aNote that 3 μ M of compound 10 produces a significant 12 mV negative shift in the half activation potential of hCav3.2 and although there is a trend for both compounds shift the half inactivation potential of the channel, it did not reach statistical significance.

of 3 μ M. This concentration blocked hCav3.2 by 69.3 \pm 4% (n = 8), which was significantly (P < 0.05) greater than that of either hCav3.1 (44.5 \pm 7%; n = 5) or hCav3.3 (42.5 \pm 5%; n = 5). Compound 9 was thus chosen for further testing in animal models of pain.

Effects of Compound 9 in Vivo on Acute Pain. Given the Cav3.2 channel blocking property of compound 9, we hypothesized that this compound may affect pain transmission in animal models. Compound 9 was delivered by either intrathecal (i.t.) or intraperitoneal (i.p.) routes, and its effects on both the acute nociceptive and the slower inflammatory pain phases of the formalin test were evaluated. One-way ANOVA revealed that i.t. treatment of mice with compound 9 (1–10 μ g/i.t., 20 min before) significantly decreased pain response time in both first (Figure 3A) and second (Figure 3B) phases (61 \pm 8% and 76 \pm 10% inhibition, respectively). I.p. treatment of mice with compound 9 (10–100 mg/kg, i.p., 30 min prior) also resulted in significantly (one-way ANOVA) reduced pain response time in both the first (Figure 3C) and second (Figure 3D) phases (47 \pm 2% and 66 \pm 48% inhibition, respectively). Importantly, systemic (via i.p.) treatment with compound 9 (30 mg/kg, i.p.) did not affect locomotor activity of mice assessed

via an open-field test (Figure 4A), suggesting that the reduced response times observed in the previous formalin tests were not due to altered motor behavior. In order to investigate if the effects observed for compound 9 were specifically mediated via T-type channels, we performed a formalin test in Cav3.2 null mice that were treated either with vehicle or with compound 9 (10 μ g/i.t.). The Cav3.2 null mice exhibited a lower mean response time when compared to wild-type mice, which is in agreement with previous data.^{14,16} As indicated in Figure 4B and C, they appear to be completely insensitive to i.t. treatment with compound 9 (10 μ g i.t.) as revealed by two-way ANOVA, indicating that compound 9 mediates its analgesic effects specifically via Cav3.2 channels.

Effect of Compound 9 on Chronic Neuropathic Pain.

To verify whether compound 9 modulates pain transmission under neuropathic conditions, we analyzed mechanical withdrawal thresholds of mice with a partial sciatic nerve injury (PNI) and treated with compound 9 (30 mg/kg, i.p.) 14 days after nerve injury. As shown in Figure 5, sciatic nerve injury triggers mechanical hyperalgesia as confirmed by significant decrease of mechanical withdrawal thresholds when compared to baselines levels (Pre-PNI, P < 0.001). Two-way ANOVA revealed that systemic (i.p.) treatment of mice with compound 9 (30 mg/kg, i.p.) significantly attenuated the mechanical hyperalgesia induced by sciatic nerve injury when compared with the PNI + Control group for longer than 3 h after treatment. These data indicate that compound 9 treatment modulates pain transmission and mediates analgesia in this animal model of chronic neuropathic pain.

DISCUSSION

T-type calcium channels are important contributors to a range of physiological functions^{2–5,32,39–41} and it is well established that the Cav3.2 channel isoform plays important roles in the afferent pain pathway.^{12,14,15,19,42–44} Finding specific and selective blockers of these channels has proven difficult as many of the well-known T-type channel blockers such as mibefradil or ethosuximide block other channels, which can then result in side effects.^{12,31,45} In this study, we developed novel compounds with structures similar to some of the endogenous ligands that are known to interact with members of the T-type channel family^{16,23,25,34,35} and then modified them to reduce their affinity for cannabinoid receptors while attempting to increase affinity for Cav3.2 channels.

We had previously synthesized a series of compounds based on endogenous cannabinoid ligands that targeted both CB receptors and T-type calcium channels.^{16,34} Using data obtained from these experiments, we designed additional compounds with an extra substituted tertiary amine attached to the carbazole scaffold. We then extended the chain length attached to the carbazole to one of the compounds (compound 9) to improve its selectivity for Cav3.2 channels over CB receptors.^{16,34} These data show that the length of the linker between the carbazole scaffold and the heterocyclic moiety is a key drug structural determinant that can be exploited to produce better and more selective Cav3.2 channel inhibitors. This compound blocked Cav3.2 channels with approximately 2-fold higher affinity than Cav3.1 and Cav3.3 channels. At this point we do not know if compound 9 affects other molecular targets such as high voltage activated channels or sodium channels. Nonetheless, the observation that their analgesic actions were abolished upon removal of Cav3.2 channels indicates that the primary biological target for compound 9 in

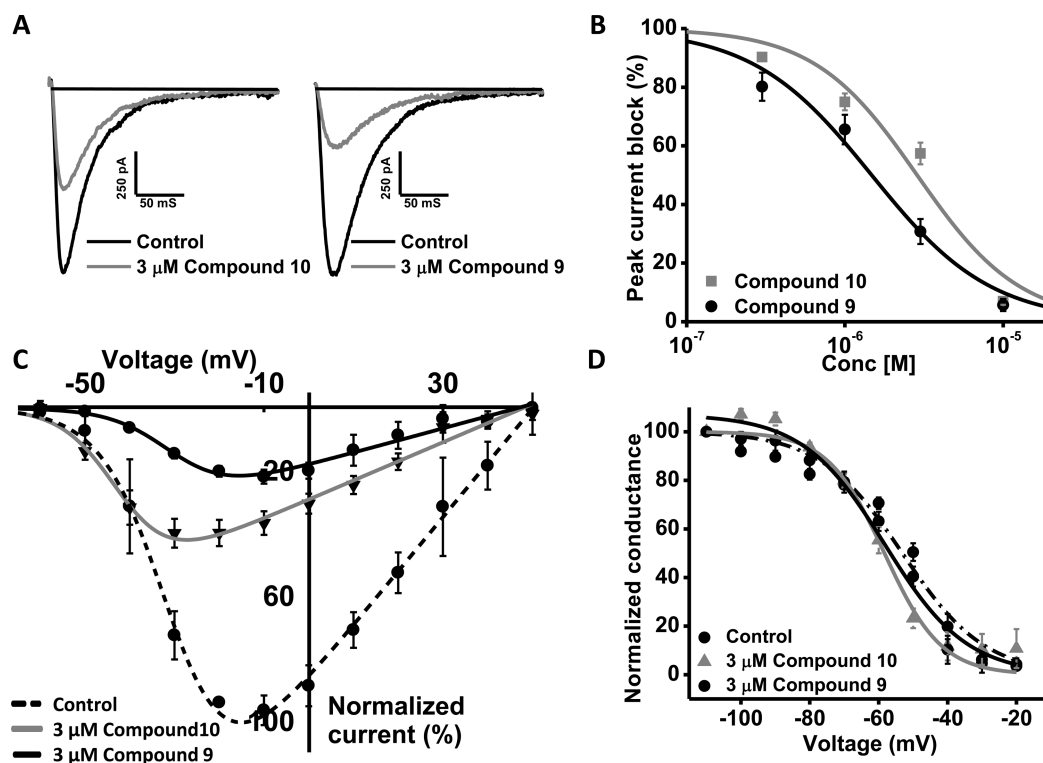


Figure 2. (A) Representative traces of hCav 3.2 before and after application of 3 μ M compounds 10 and 9. (B) Dose–response relations for compound 9 and 10 block of hCav3.2 channels. The IC_{50} from the fit with the Hill equation was 1.48 and 3.68 μ M, respectively ($n = 6$). (C) Effect of 3 μ M compounds 9 and 10 on the steady state inactivation curve for Cav3.2 channels. (D) Effect of 3 μ M compounds 9 and 10 on the current voltage relation for Cav3.2 channels. Note: Data in panels (B) and (C) were fitted with the Boltzmann equation, and data were obtained from 6 paired experiments.

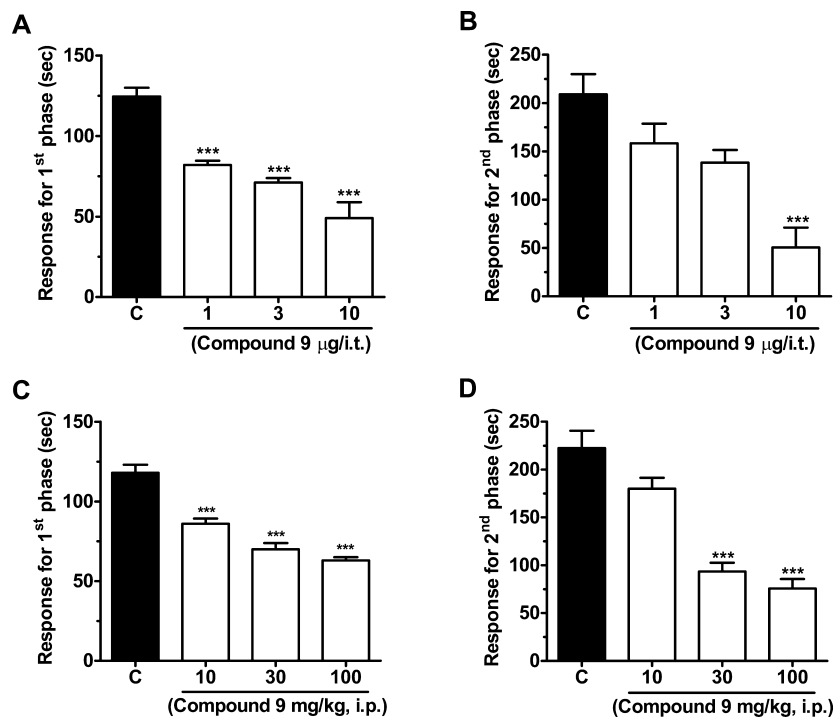


Figure 3. (A, B) Effect of increasing doses of intrathecal compound 9 on the first and second phases of formalin-induced pain. (C, D) Effect of increasing doses of intraperitoneal compound 9 (on the first and second phases of formalin-induced pain. Each bar represents the mean \pm SEM ($n = 6-8$), and is representative of 2 independent experiments. Asterisks denote the significance relative to the control group (***) $P < 0.001$, one-way ANOVA followed by Dunnett's test).

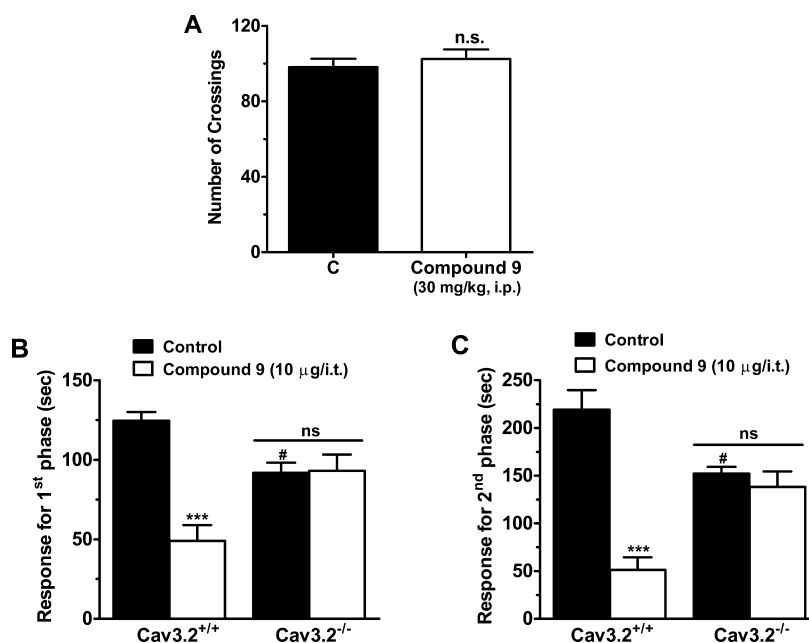


Figure 4. (A) Effect of 30 mg/kg intraperitoneal compound 9 on locomotor activity of wild type mice in the open field test. (B, C) Comparison of effect of 10 $\mu\text{g}/\text{i.t.}$ intrathecal compound 9 on the first and second phases of formalin-induced pain in wild type and Cav3.2 knockout mice, respectively. Each bar represents the mean \pm SEM ($n = 6-7$) and is representative of 2 independent experiments. Asterisks denote the significance relative to the control group $***P < 0.001$ when comparing treatment; and $\#P < 0.05$, for comparison between genotypes (two-way ANOVA followed by Tukey's test). Note that control mice were of the same genetic background as the Cav3.2 null mice.

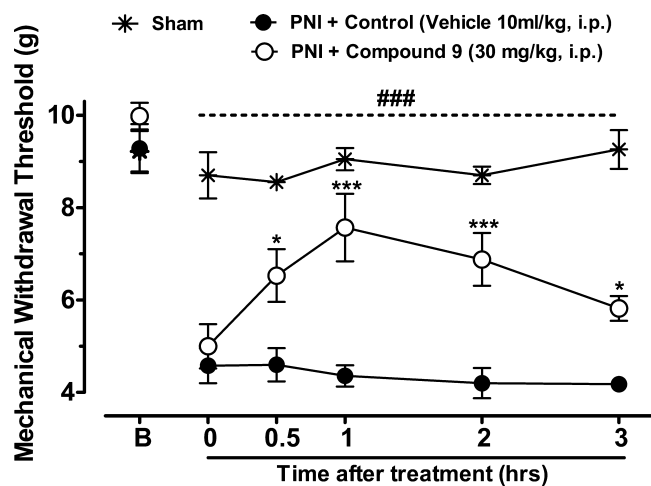


Figure 5. Blind analyses of the time course of treatment of neuropathic mice with vehicle or compound 9. Each circle represents the mean \pm SEM ($n = 6$), and is representative of 2 independent experiments. ($*P < 0.05$, $***P < 0.001$, two-way ANOVA followed by Tukey's test). The dashed line and number symbols indicate the range of data points where injured animals significantly differed from the sham treated group ($P < 0.001$).

the context of pain signaling is Cav3.2. The potent effects of compound 9 on pain response in injured wild type animals fits with the notion that Cav3.2 channels play an important role in the afferent pain pathway^{17,18,20,21,42,44,46,47} and also with a number of previous studies showing that Cav3.2 channel blockers are efficacious in various pain models.^{12,15,20,29} Calculated physicochemical properties such as topological polar surface area (tPSA), lipophilicity (clogP), and number of hydrogen bond donor and acceptor atoms are useful indicator of druglike properties. Poor oral absorption is expected for compounds with a TPSA more than 140 \AA^2 ,

with a log $D > 5$, more than 5 hydrogen bond donors, and more than 10 hydrogen bond acceptors. Compound 9 has a TPSA value of 66.37 \AA^2 , a molecular weight of 490.68 g/mol, a clogP value of 4.59, and less than five hydrogen bond acceptors or donors, suggesting a reasonable probability of good oral absorption and intestinal permeability.

Compound 9 blocked Cav3.2 channels in a concentration range that was similar to that previously reported for compounds such as NMP7³⁴ and NMP181.¹⁶ These two molecules, although structurally related to compound 9 did not discriminate among the three Cav3 family members, and mediated leftward shifts in the midpoint of the steady state inactivation curve. In contrast, compound 9 did not significantly affect voltage-dependent inactivation properties of Cav3.2 channels, suggesting that this compound does not interact strongly with inactivated channels. At this point, it is not clear which structural differences among these compounds are responsible for these differential effects on channel gating. While other types of T-type calcium channel blockers such as TTA-2⁴⁸ or Z123212²⁸ that are efficacious in various pain models are also known to promote voltage-dependent inactivation of Cav3.2 channels, our data showing efficacy of compound 9 in neuropathic and inflammatory pain indicate that such state-dependent inhibition is not a prerequisite for antihyperalgesic effects in vivo.

Altogether our data suggest that using a carbazole scaffold is an effective strategy for developing potent Cav3.2 calcium channel blockers for therapeutic intervention into inflammatory and neuropathic pain hypersensitivity. In addition, T-type channels are also associated with many other disorders, including epilepsy and cardiac hypertrophy;^{2-4,7-11,18,32,49} therefore, the novel pharmacophores that we have identified here may prove useful toward treatment of these disorders.

METHODS

In Vitro Receptor Radioligand CB₁ and CB₂ Binding Studies.

CB₁ and CB₂ radioligand binding data were obtained using National Institute of Mental Health (NIMH) Psychoactive Drug Screening Program (PDSP) resources as described earlier.^{34,50–52}

cDNA Constructs. Human Cav3.2 cDNA construct was kindly provided by Dr. Terrance Snutch (University of British Columbia, Vancouver, Canada).

tsA-201 Cell Culture and Transfection. Human embryonic kidney tsA-201 cells were cultured and transfected using the calcium phosphate method as described previously.⁵³ Transfected cells were then incubated 48 h at 37 °C and 5% CO₂ and then resuspended with 0.25% (w/v) trypsin-EDTA (Invitrogen) and plated on glass coverslips a minimum of 3 to 4 h before patching.

Electrophysiology. Whole-cell voltage-clamp recordings from tsA-201 cells were performed at room temperature 2–3 days after transfection. External recording solution contained (in mM): 114 CsCl, 20 BaCl₂, 1 MgCl₂, 10 HEPES, 10 glucose, adjusted to pH 7.4 with CsOH. Internal patch pipet solution contained the following (in mM): 126.5 CsMeSO₄, 2 MgCl₂, 11 EGTA, 10 HEPES adjusted to pH 7.3 with CsOH. Internal solution was supplemented with 0.6 mM GTP and 2 mM ATP and mixed thoroughly just prior to use. Liquid junction potentials for the recording solutions were left uncorrected.

Tested compounds were prepared daily from DMSO stocks diluted in external solution. Using a custom built gravity driven micro-perfusion system,⁵³ diluted compounds were then applied rapidly and locally to the cells. Control vehicle experiments were performed to ensure that 0.1% DMSO had no effect on current amplitudes or on the half a ctivation and half-inactivation potentials (data not shown). Currents were measured using the whole-cell patch clamp technique and an Axopatch 200B amplifier in combination with Clampex 9.2 software (Molecular Devices, Sunnyvale, CA). After establishing whole cell configuration, cellular capacitance was minimized using the amplifier's built-in analog compensation. Series resistance was compensated by at least 85% in all experiments. All data were digitized at 10 kHz with a Digidata 1320 interface (Molecular Devices) and filtered at 1 kHz (8-pole Bessel filter). Raw and online leak-subtracted data were both collected simultaneously. In current–voltage relation studies, the membrane potential was held at –110 mV and cells were depolarized from –80 to 20 mV in 10 mV increments. For steady-state inactivation studies, a 3.6 s conditioning prepulse of various magnitude (initial holding at –110 mV) was followed by a depolarizing pulse to –20 mV. Individual sweeps were separated by 12 s to permit recovery from inactivation between conditioning pulses. The duration of the test pulse was typically 200 ms and the current amplitude obtained from each test pulse was normalized to that observed at the holding potential of –110 mV.

Animals. During experiments, all efforts were made to minimize animal suffering according to the policies and recommendations of the International Association for the Study of Pain and all protocols used were approved by the Institutional Animal Care and Use Committee. For all experiments, either adult male C57BL/6J (wild-type) or CACNA1H knockout (Cav3.2 null) mice (20–25g) purchased from Jackson Laboratories were used. There were a maximum of five mice per cage (30 × 20 × 15 cm²), and access to food and water was unlimited. Temperature was kept at 23 ± 1 °C on a 12 h light/dark cycle (lights on at 7:00 a.m.). Intraperitoneal (i.p.) injections of drugs were a constant volume of 10 mL/kg body weight. Intrathecal (i.t.) injections used volumes of 10 μL and were performed using the method described previously⁵⁴ and carried out routinely in our laboratory.^{16,55} All drugs were dissolved in 1% or less DMSO, whereas control animals received PBS + 1% DMSO. For each test, a different group of mice were used and only one experiment per mouse was performed. In experiments examining the action of compound 9 on PNI-induced neuropathy, the observer was blind to the conditions tested.

Formalin Test. Before experiments were performed, mice were left to acclimatize for at least 60 min. Animals were then injected intraplantarly (i.pl.) in the ventral surface of the right hindpaw with 20

μL of a formalin solution (1.25%) made up in PBS. Following i.pl. injections of formalin, individual animals were placed immediately into observation chambers and monitored from 0 to 5 min (acute nociceptive phase) and 15–30 min (inflammatory phase). The time spent licking or biting the injected paw was then considered as a nociceptive response and recorded with a chronometer. Compound 9 was delivered by i.t. (20 min prior) or by i.p. (30 min prior) and its effect on both the nociceptive and inflammatory phases of the formalin test was evaluated.

Open-Field Test. Animals received compound 9 via i.p. (30 mg/kg) route 30 min before testing, and the ambulatory behavior of the treated animals was assessed in an open-field test as described previously.⁵⁶ Briefly, the apparatus consisted of a wooden box measuring 40 × 60 × 50 cm³ with a glass front wall. The floor was divided into 12 equal squares and the entire apparatus was placed in a sound free room. Animals were placed in the rear left square and left to explore freely and the number of squares crossed with all paws (crossing) in a 6 min time frame was counted. After each individual mouse session, the apparatus was then cleaned and dried with a 10% alcohol solution.

Partial Sciatic Nerve Injury (PNI)-Induced Neuropathic Pain. Before surgery, mice were anaesthetized with isoflurane (5% induction, 2.5% maintenance). Partial ligation of the sciatic nerve was performed by tying the distal 1/3 to 1/2 of the dorsal portion as previously described.⁵⁷ In sham-operated mice, the sciatic nerve was exposed without ligation and all wounds were closed and treated with iodine solution. After 14 days post surgery, mice received either compound 9 (30 mg/kg, i.p.) treatment or vehicle, while sham-operated animals received only vehicle (10 mL/kg, i.p.). Mechanical hyperalgesia was then evaluated in a time-dependent manner.

Evaluation of Mechanical Hyperalgesia. Mechanical hyperalgesia responses were recorded immediately before the surgeries (baselines), 14 days after the surgeries (0), and at various time points (0.5, 1, 2, 3 h) after treatment. Measurements were made using a Dynamic Plantar Aesthesiometer (DPA, Ugo Basile, Varese, Italy). Briefly, individual animals were placed in small enclosed testing arenas (20 cm × 18.5 cm × 13 cm, length × width × height) on top of a wire mesh floor and allowed to acclimate for a period of at least 90 min. The DPA device was then positioned so that the filament was directly under the plantar surface of the ipsilateral hind paw of the animal and tested three times per session.

Data Analysis and Statistics. Data were analyzed using Clampfit 9.2 (Molecular Devices). Origin 7.5 software (Northampton, MA) was used in the preparation of all figures and curve fittings. Current–voltage relationships were fitted with the modified Boltzmann equation: $I = [G_{\max}(V_m - E_{\text{rev}})]/[1 + \exp((V_{0.5\text{sact}} - V_m)/k_a)]$, where V_m is the test potential, $V_{0.5\text{sact}}$ is the half activation potential, E_{rev} is the reversal potential, G_{\max} is the maximum slope conductance, and k_a reflects the slope of the activation curve. Steady-state inactivation curves were fitted using the Boltzmann equation: $I = 1/(1 + \exp((V_m - V_h)/k))$, where V_h is the half-inactivation potential and k is the slope factor. Dose–response curves were fitted with the equation $y = A_2 + (A_1 - A_2)/(1 + ([C]/IC_{50})^n)$, where A_1 is initial current amplitude and A_2 is the current amplitude at saturating drug concentrations, $[C]$ is the drug concentration, and n is the Hill coefficient. Statistical significance was determined by paired or unpaired Student's *t* tests and one-way or repeated measures ANOVA, followed by Dunnett's test or Tukey's multiple comparison tests. Significant values were set as indicated in the text and figure legends. All data are given as means ± standard errors.

Chemistry. All moisture sensitive reactions were performed in an inert, dry atmosphere of argon in flame-dried glassware. Air sensitive liquids were transferred via syringe or cannula through rubber septa. Reagent grade solvents were used for extraction and flash chromatography. THF was distilled from Na/benzophenone under argon; dichloromethane (CH₂Cl₂) and chloroform (CHCl₃) were distilled from CaH₂ under argon. All other reagents and solvents which were purchased from commercial sources were used directly without further purification. The progress of reactions was checked by analytical thin-layer chromatography (Sorbent Technologies, Silica G

TLC plates w/UV 254). The plates were visualized first with UV illumination followed by charring with ninhydrin (0.3% ninhydrin (w/v), 97:3 EtOH-AcOH). Flash column chromatography was performed using prepacked Biotage SNAP cartridges on a Biotage Isolera One instrument. The solvent compositions reported for all chromatographic separations are on a volume/volume (v/v) basis. ¹H NMR spectra were recorded at 400 or 500 MHz and are reported in parts per million (ppm) on the δ scale relative to tetramethylsilane as an internal standard. ¹³C NMR spectra were recorded at 100 or 125 MHz and are reported in parts per million (ppm) on the δ scale relative to CDCl₃ (δ 77.00). Melting points were determined on a Stuart melting point apparatus from Bibby Scientific Limited and are uncorrected. High Resolution mass spectrometry (HRMS) was performed on a Waters/Micromass LCT-TOF instrument. All compounds were more than 95% pure.

N-((1-(2-(*tert*-Butylamino)-2-oxoethyl)piperidin-4-yl)methyl)-9-pentyl-9*H*-carbazole-3-carboxamide (**9**). Under nitrogen atmosphere, a mixture of 9-pentyl-*N*-(piperidin-4-ylmethyl)-9*H*-carbazole-3-carboxamide **7**¹⁶ (389 mg, 1.03 mmol), *N*-*tert*-butyl-2-chloroacetamide (185 mg, 1.24 mmol), potassium carbonate (427 mg, 3.09 mmol), and potassium iodide (160 mg, 0.96 mmol) in *n*-butanol (5 mL) was subjected to microwave irradiation at 110 °C for 3 h. The mixture was allowed to cool to room temperature, diluted with DCM (50 mL), and filtered. The organic solvents were evaporated in vacuo. The residue was purified on a Biotage KP-NH cartridge (amino-modified silica gel) using cyclohexane/EtOAc in different proportions to afford the title compound as a light yellow glass. Yield: 476 mg (94%). ¹H NMR (400 MHz, CDCl₃) δ 8.49 (d, *J* = 1.4 Hz, 1H), 8.01 (d, *J* = 7.9 Hz, 1H), 7.84 (dd, *J* = 8.6, 1.4 Hz, 1H), 7.41 (ddd, *J* = 8.3, 7.1, 1.1 Hz, 1H), 7.33 (d, *J* = 8.3 Hz, 1H), 7.29 (d, *J* = 8.6 Hz, 1H), 7.16 (ddd, *J* = 7.9, 7.1, 1.1 Hz, 1H), 7.00 (br s, 1H), 6.58 (t, *J* = 5.8 Hz, 1H), 4.19 (t, *J* = 7.2 Hz, 2H), 3.34 (t, *J* = 6.5 Hz, 2H), 2.81–2.67 (m, 4H), 2.00 (td, *J* = 11.7, 2.5 Hz, 2H), 1.83–1.74 (m, 2H), 1.71 (br d, *J* = 13.9 Hz, 2H), 1.64–1.48 (m, 1H), 1.31–1.22 (m, 15H), 0.78 (t, *J* = 6.9 Hz, 3H). ¹³C and DEPT NMR (101 MHz, CDCl₃) δ 169.84 (C=O), 168.52 (C=O), 142.33 (C), 141.15 (C), 126.44 (CH), 125.28 (C), 124.81 (CH), 122.96 (C), 122.73 (C), 120.68 (CH), 119.85 (CH), 119.68 (CH), 109.23 (CH), 108.50 (CH), 62.73 (CH₂), 53.89 (CH₂), 50.51 (C), 45.57 (CH₂), 43.37 (CH₂), 36.23 (CH), 30.61 (CH₂), 29.48 (CH₂), 28.93 (CH₃), 28.76 (CH₂), 22.57 (CH₂), 14.08 (CH₃). ESI: *m/z* 491.2 (M + H)⁺. HRMS calcd for C₃₀H₄₃N₄O₂ (M + H)⁺ 491.3386, found 491.3433.

N-*tert*-Butyl-2-[4-(9-pentyl-9*H*-carbazole-3-carbonyl)piperazin-1-yl]acetamide (**10**). Under argon atmosphere, to a solution of *tert*-butyl 4-(9-pentyl-9*H*-carbazole-3-carbonyl)piperazine-1-carboxylate **8**¹⁶ (360 mg, 1.03 mmol) and MeCN (10 mL) was added K₂CO₃ (427 mg, 3.09 mmol), KI (160 mg, 0.96 mmol), and *N*-*tert*-butyl-2-chloroacetamide (185 mg, 1.24 mmol). The reaction mixture was stirred at reflux for 3 h. The reaction mixture was cooled, diluted with DCM (30 mL), and filtered. The organic solvents were evaporated in vacuo. The organic layer was washed three times with brine (50 mL). The organic layer was then separated, dried over magnesium sulfate, filtered, and concentrated in vacuo. The crude product was purified by flash chromatography eluting with heptane/EtOAc (0–100%) to yield the title compound (437 mg, 92%) as yellow foam. ¹H NMR (400 MHz, CDCl₃) δ 8.19 (d, *J* = 1.2 Hz, 1H), 8.09 (d, *J* = 7.6 Hz, 1H), 7.38–7.55 (m, 4H), 7.23–7.28 (m, 1H), 6.96 (s, 1H), 4.30 (t, *J* = 7.2 Hz, 2H), 3.74 (br s, 4H), 2.96 (s, 2H), 2.58 (br s, 4H), 1.86 (br s, 2H), 1.24–1.39 (m, 13H), 0.81–1.01 (m, 3H). ¹³C NMR (101 MHz, CDCl₃) δ ppm 171.69 (C=O), 168.71 (C=O), 141.20 (C), 140.94 (C), 126.30 (CH), 125.60 (C), 125.19 (CH), 122.61 (C), 122.50 (C), 120.54 (CH), 120.14 (CH), 119.43 (CH), 109.07 (CH), 108.51 (CH), 77.44 (CH₂), 77.33 (CH₂), 77.13 (CH₂), 76.81 (CH₂), 62.27 (CH₂), 53.48 (CH₂), 50.64 (C), 43.24 (CH₂), 29.38 (CH₂), 28.83 (CH₂), 28.66 (CH₃), 22.48 (CH₂), 13.98 (CH₃). *m/z* 463.2 (M + H)⁺. HRMS calcd for C₃₀H₄₃N₄O₂ (M + H)⁺ 463.3073, found 463.2819.

tert-Butyl-4-((9-Propyl-9*H*-carbazole-3-carboxamido)methyl)piperidine-1-carboxylate (**11**). 9-Propyl-9*H*-carbazole-3-carboxylic acid¹⁶ (100 mg, 0.40 mmol), HOBt (62.32 mg, 0.48 mmol), DIPEA

(0.13 mL, 0.79 mmol), DMAP (58.12 mg, 0.40 mmol), and EDAC (91.22 mg, 0.40 mmol) were added upon stirring to DCM (9 mL) under nitrogen. The obtained solution was cooled down on an ice-water bath, *tert*-butyl 4-(aminomethyl)piperidine-1-carboxylate (119.24 mg, 0.40 mmol) was added in one portion, and the resulting reaction mixture was then allowed to warm to room temperature and stirred for 16 h. The solvent was removed in vacuo, and the obtained residue was extracted with EtOAc (50 mL). The organic layer was washed three times with brine (25 mL). The organic layer was then separated, dried over magnesium sulfate, filtered, and concentrated in vacuo. The crude product was purified by flash chromatography eluting with heptane/EtOAc (0–100%) to produce the title compound **11** (158.3 mg, 89%) as yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.57 (d, *J* = 1.4 Hz, 1H), 8.06 (d, *J* = 8.0 Hz, 1H), 7.89 (dd, *J* = 8.7, 1.4 Hz, 1H), 7.44 (ddd, *J* = 8.3, 7.1, 1.1 Hz, 1H), 7.31 (d, *J* = 8.3 Hz, 1H), 7.27 (d, *J* = 8.7 Hz, 1H), 7.19 (ddd, *J* = 8.0, 7.1, 1.1 Hz, 1H), 6.55 (t, *J* = 5.8 Hz, 1H), 4.26 (t, *J* = 7.2 Hz, 2H), 3.38 (t, *J* = 6.2 Hz, 2H), 3.10 (d, *J* = 12.1 Hz, 2H), 2.61 (td, *J* = 12.2, 2.5 Hz, 2H), 1.99 (s, 1H), 1.97–1.80 (m, 2H), 1.80–1.70 (m, 3H), 1.45 (s, 9H), 1.22 (qd, *J* = 12.8, 3.7 Hz, 2H), 0.97 (t, *J* = 7.4 Hz, 3H). ¹³C (101 MHz, CDCl₃) δ 168.53, 154.77, 142.30, 141.16, 126.42, 125.42, 124.77, 122.97, 122.74, 120.74, 119.83, 119.68, 109.22, 108.53, 79.22, 45.22, 45.39, 36.91, 31.38, 28.41, 22.58, 12.10.

9-Propyl-*N*-(piperidin-4-ylmethyl)-9*H*-carbazole-3-carboxamide (**12**). Trifluoroacetic acid (5 mL) was added to a solution of **11** (150 mg, 0.33 mmol) in CH₂Cl₂ (5 mL), and the reaction mixture was stirred at room temperature for 4 h. The mixture was evaporated, and the residue was basified with 2 N NaOH. The solution was extracted with CH₂Cl₂ three times. The extracts were dried with Na₂SO₄ and evaporated. The crude product was purified by flash chromatography eluting with heptane/EtOAc (0–100%) to produce the title compound **12** (95.5 mg, 83%) as dark yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.56 (d, *J* = 1.5 Hz, 1H), 8.05 (d, *J* = 8.0 Hz, 1H), 7.90 (dd, *J* = 8.7, 1.5 Hz, 1H), 7.5 (ddd, *J* = 8.3, 7.2, 1.2 Hz, 1H), 7.31 (d, *J* = 8.3 Hz, 1H), 7.28 (d, *J* = 8.7 Hz, 1H), 7.17 (ddd, *J* = 8.0, 7.2, 1.2 Hz, 1H), 6.58 (t, *J* = 5.8 Hz, 1H), 4.25 (t, *J* = 7.1 Hz, 2H), 3.39 (t, *J* = 6.2 Hz, 2H), 3.09 (d, *J* = 12.1 Hz, 2H), 2.60 (td, *J* = 12.2, 2.5 Hz, 2H), 1.99 (s, 1H), 1.98–1.80 (m, 2H), 1.80–1.71 (m, 3H), 1.23 (qd, *J* = 12.8, 3.7 Hz, 2H), 0.96 (t, *J* = 7.5 Hz, 3H). ¹³C (101 MHz, CDCl₃) δ 168.51, 142.32, 141.15, 126.40, 125.41, 124.79, 122.99, 122.73, 120.73, 119.82, 119.67, 109.21, 108.50, 46.47, 46.20, 45.37, 36.91, 31.37, 22.59, 12.10.

N-((1-(2-(*tert*-Butylamino)-2-oxoethyl)piperidin-4-yl)methyl)-9-propyl-9*H*-carbazole-3-carboxamide (**13**). Under argon atmosphere, to a solution of **12** (90 mg, 0.26 mmol) and MeCN (2.5 mL) was added K₂CO₃ (107.8 mg, 0.78 mmol), KI (43.2 mg, 0.26 mmol), and *N*-*tert*-butyl-2-chloroacetamide (47.9 mg, 0.32 mmol). The title compound was prepared according to the analogous procedure described above for compound **9** to produce the title compound **13** (121.3 mg, 82%) as dark yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.56 (d, *J* = 1.5 Hz, 1H), 8.05 (d, *J* = 8.0 Hz, 1H), 7.91 (dd, *J* = 8.7, 1.5 Hz, 1H), 7.44 (ddd, *J* = 8.2, 7.1, 1.1 Hz, 1H), 7.31 (d, *J* = 8.2 Hz, 1H), 7.28 (d, *J* = 8.7 Hz, 1H), 7.17 (ddd, *J* = 8.0, 7.1, 1.1 Hz, 1H), 6.98 (br s, 1H), 6.58 (t, *J* = 5.8 Hz, 1H), 4.26 (t, *J* = 7.1 Hz, 2H), 3.33 (t, *J* = 6.3 Hz, 2H), 2.82–2.67 (m, 4H), 2.03 (td, *J* = 11.8, 2.7 Hz, 2H), 1.86–1.76 (m, 2H), 1.7 (d, *J* = 13.7 Hz, 2H), 1.64–1.48 (m, 1H), 1.34 (s, 9H), 0.94 (t, *J* = 7.0 Hz, 3H). ¹³C and DEPT NMR (101 MHz, CDCl₃) δ 169.84 (C=O), 168.5 (C=O), 142.34 (C), 141.17 (C), 126.41 (CH), 126.0 (C), 124.79 (CH), 122.94 (C), 122.71 (C), 120.68 (CH), 119.84 (CH), 119.69 (CH), 109.23 (CH), 108.51 (CH), 62.74 (CH₂), 53.9 (CH₂), 50.52 (C), 45.58 (CH₂), 45.35 (CH₂), 36.22 (CH), 30.61 (CH₂), 28.90 (CH₃), 22.76 (CH₂), 12.18 (CH₃). ESI: *m/z* 463.3 (M + H)⁺. HRMS calcd for C₂₈H₄₀N₄O₂ (M + H)⁺ 463.3073, found 463.3089.

tert-Butyl 4-((9-Butyl-9*H*-carbazole-3-carboxamido)methyl)piperidine-1-carboxylate (**14**). Using 9-butyl-9*H*-carbazole-3-carboxylic acid¹⁶ (100 mg, 0.37 mmol), HOBt (60.54 mg, 0.45 mmol), DIPEA (0.13 mL, 0.75 mmol), DMAP (54.73 mg, 0.45 mmol), *tert*-butyl 4-(aminomethyl)piperidine-1-carboxylate (112.22 mg, 0.45 mmol), and EDAC (85.85 mg, 0.45 mmol), the title compound was

prepared according to the analogous procedure described above for compound **11**, to produce the title compound **14** (147.3 mg, 85%) as a yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.53 (d, $J = 1.5$ Hz, 1H), 8.04 (d, $J = 7.9$ Hz, 1H), 7.82 (dd, $J = 8.6, 1.5$ Hz, 1H), 7.43 (ddd, $J = 8.3, 7.1, 1.1$ Hz, 1H), 7.31 (d, $J = 8.3$ Hz, 1H), 7.27 (d, $J = 8.6$ Hz, 1H), 7.16 (ddd, $J = 7.9, 7.1, 1.1$ Hz, 1H), 6.57 (t, $J = 5.8$ Hz, 1H), 4.24 (t, $J = 7.2$ Hz, 2H), 3.37 (t, $J = 6.2$ Hz, 2H), 3.07 (d, $J = 12.1$ Hz, 2H), 2.61 (td, $J = 12.2, 2.5$ Hz, 2H), 2.00 (s, 1H), 1.98–1.80 (m, 2H), 1.80–1.71 (m, 3H), 1.45 (s, 9H), 1.43–1.32 (m, 2H), 1.23 (qd, $J = 12.8, 3.7$ Hz, 2H), 0.90 (t, $J = 7.4$ Hz, 3H). ^{13}C (101 MHz, CDCl_3) δ 168.50, 154.75, 142.33, 141.16, 126.45, 125.43, 124.80, 122.97, 122.72, 120.73, 119.81, 119.68, 109.21, 108.50, 79.26, 45.47, 43.04, 36.90, 31.36, 28.41, 20.78, 14.13.

9-Butyl-N-(piperidin-4-ylmethyl)-9H-carbazole-3-carboxamide (15). Trifluoroacetic acid (5 mL) was added to a solution of **14** (140 mg, 0.30 mmol) in CH_2Cl_2 (5 mL), and the title compound was prepared according to the analogous procedure described above for compound **12** to produce the title compound **15** (95.7 mg, 88%) as dark yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.53 (d, $J = 1.5$ Hz, 1H), 8.02 (d, $J = 7.9$ Hz, 1H), 7.84 (dd, $J = 8.7, 1.5$ Hz, 1H), 7.41 (ddd, $J = 8.3, 7.2, 1.1$ Hz, 1H), 7.30 (d, $J = 8.3$ Hz, 1H), 7.27 (d, $J = 8.7$ Hz, 1H), 7.17 (ddd, $J = 7.9, 7.2, 1.1$ Hz, 1H), 6.58 (t, $J = 5.8$ Hz, 1H), 4.25 (t, $J = 7.2$ Hz, 2H), 3.39 (t, $J = 6.2$ Hz, 2H), 3.09 (d, $J = 12.2$ Hz, 2H), 2.60 (td, $J = 12.2, 2.5$ Hz, 2H), 1.99 (s, 1H), 1.98–1.80 (m, 2H), 1.80–1.71 (m, 3H), 1.43–1.32 (m, 2H), 1.23 (qd, $J = 12.8, 3.7$ Hz, 2H), 0.90 (t, $J = 7.4$ Hz, 3H). ^{13}C (101 MHz, CDCl_3) δ 168.51, 142.32, 141.15, 126.45, 125.41, 124.79, 122.99, 122.73, 120.73, 119.82, 119.67, 109.21, 108.50, 46.47, 46.20, 43.03, 36.91, 31.37, 20.79, 14.12.

N-((1-(2-(tert-Butylamino)-2-oxoethyl)piperidin-4-yl)methyl)-9-butyl-9H-carbazole-3-carboxamide (16). Under argon atmosphere, to a solution of **15** (90 mg, 0.25 mmol) and MeCN (2.5 mL) was added K_2CO_3 (103.7 mg, 0.75 mmol), KI (41.5 mg, 0.25 mmol), and *N*-tert-butyl-2-chloroacetamide (44.9 mg, 0.30 mmol). The title compound was prepared according to the analogous procedure described above for compound **9** to produce the title compound **16** (99.9 mg, 84%) as dark yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.50 (d, $J = 1.5$ Hz, 1H), 8.01 (d, $J = 7.9$ Hz, 1H), 7.8 (dd, $J = 8.6, 1.5$ Hz, 1H), 7.4 (ddd, $J = 8.3, 7.1, 1.1$ Hz, 1H), 7.30 (d, $J = 8.3$ Hz, 1H), 7.27 (d, $J = 8.6$ Hz, 1H), 7.16 (ddd, $J = 7.9, 7.1, 1.1$ Hz, 1H), 7.01 (br s, 1H), 6.57 (t, $J = 5.8$ Hz, 1H), 4.24 (t, $J = 7.2$ Hz, 2H), 3.35 (t, $J = 6.5$ Hz, 2H), 2.83–2.68 (m, 4H), 2.00 (td, $J = 11.7, 2.5$ Hz, 2H), 1.82–1.72 (m, 2H), 1.73 (d, $J = 13.8$ Hz, 2H), 1.63–1.49 (m, 1H), 1.45–1.34 (m, 2H), 1.33 (s, 9H), 0.90 (t, $J = 6.8$ Hz, 3H). ^{13}C and DEPT NMR (101 MHz, CDCl_3) δ 169.83 (C=O), 168.57 (C=O), 142.34 (C), 141.14 (C), 126.45 (CH), 125.26 (C), 124.82 (CH), 122.94 (C), 122.73 (C), 120.68 (CH), 119.85 (CH), 119.68 (CH), 109.21 (CH), 108.50 (CH), 62.71 (CH₂), 53.90 (CH₂), 50.50 (C), 45.59 (CH₂), 43.00 (CH₂), 36.26 (CH), 30.66 (CH₂), 29.43 (CH₂), 28.92 (CH₃), 20.74 (CH₂), 14.10 (CH₃). ESI: m/z 477.2 (M + H)⁺. HRMS calcd for $\text{C}_{29}\text{H}_{41}\text{N}_4\text{O}_2$ (M + H)⁺ 477.3229, found 477.3217.

tert-Butyl 4-((9-Pentyl-9H-carbazole-3-carboxamido)methyl)pyrrolidine-1-carboxylate (17). Using 9-pentyl-9H-carbazole-3-carboxylic acid (100 mg, 0.36 mmol), HOBt (57.81 mg, 0.43 mmol), DIPEA (0.12 mL, 0.71 mmol), DMAP (52.29 mg, 0.43 mmol), (R)-3-(aminomethyl)-1-Boc-pyrrolidine (85.66 mg, 0.43 mmol), and EDAC (82.02 mg, 0.43 mmol), the title compound was prepared according to the analogous procedure described above for compound **11** using the amide coupling protocol, to produce the title compound **17** (141.8 mg, 86%) as a yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.58 (d, $J = 1.6$ Hz, 1H), 8.02 (d, $J = 8.0$ Hz, 1H), 7.83 (dd, $J = 8.6, 1.6$ Hz, 1H), 7.42 (ddd, $J = 8.3, 7.1, 1.1$ Hz, 1H), 7.31 (d, $J = 8.3$ Hz, 1H), 7.28 (d, $J = 8.6$ Hz, 1H), 7.17 (ddd, $J = 8.0, 7.1, 1.1$ Hz, 1H), 6.57 (t, $J = 5.8$ Hz, 1H), 4.16 (t, $J = 7.2$ Hz, 2H), 3.22–3.12 (m, 2H), 2.34–2.21 (m, 2H), 2.12–1.98 (m, 1H), 1.81–1.67 (m, 3H), 1.45 (s, 9H), 1.38–1.20 (m, 4H), 0.80 (t, $J = 6.9$ Hz, 3H). ^{13}C (101 MHz, CDCl_3) δ 168.56, 154.75, 142.32, 141.13, 126.47, 125.25, 124.82, 122.92, 122.72, 120.69, 119.85, 119.68, 109.22, 108.50, 79.27, 54.28, 45.50, 39.78, 38.75, 36.26, 30.68, 29.45, 28.42, 20.75, 14.10.

9-Pentyl-N-(pyrrolidin-4-ylmethyl)-9H-carbazole-3-carboxamide (18). Trifluoroacetic acid (5 mL) was added to a solution of

compound **17** (135 mg, 0.29 mmol) in CH_2Cl_2 (5 mL), the title compound was prepared according to the analogous procedure described above for compound **12** to produce the title compound **18** (90.7 mg, 86%) as yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.57 (d, $J = 1.5$ Hz, 1H), 8.03 (d, $J = 7.9$ Hz, 1H), 7.84 (dd, $J = 8.6, 1.5$ Hz, 1H), 7.41 (ddd, $J = 8.3, 7.1, 1.1$ Hz, 1H), 7.32 (d, $J = 8.3$ Hz, 1H), 7.27 (d, $J = 8.6$ Hz, 1H), 7.17 (ddd, $J = 7.9, 7.1, 1.1$ Hz, 1H), 6.57 (t, $J = 5.84$ Hz, 1H), 4.16 (t, $J = 7.2$ Hz, 2H), 3.22–3.13 (m, 2H), 2.34–2.22 (m, 2H), 2.12–1.99 (m, 1H), 1.82–1.68 (m, 3H), 1.38–1.20 (m, 4H), 0.82 (t, $J = 6.9$ Hz, 3H). ^{13}C (101 MHz, CDCl_3) δ 168.55, 142.33, 141.14, 126.46, 125.24, 124.82, 122.93, 122.72, 120.69, 119.85, 119.68, 109.22, 108.50, 54.27, 45.50, 39.79, 38.76, 36.26, 30.68, 29.44, 20.76, 14.11.

N-((1-(2-(tert-Butylamino)-2-oxoethyl)pyrrolidin-4-yl)methyl)-9-pentyl-9H-carbazole-3-carboxamide (19). Under argon atmosphere, to a solution of compound **18** (85 mg, 0.23 mmol) and MeCN (2.5 mL) was added K_2CO_3 (95.4 mg, 0.69 mmol), KI (38.1 mg, 0.69 mmol), and *N*-tert-butyl-2-chloroacetamide (41.9 mg, 0.28 mmol). The title compound was prepared according to the analogous procedure described above for compound **10** to produce the title compound **19** (96.2 mg, 88%) as dark yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.58 (d, $J = 1.5$ Hz, 1H), 8.02 (d, $J = 7.9$ Hz, 1H), 7.83 (dd, $J = 8.6, 1.5$ Hz, 1H), 7.41 (ddd, $J = 8.3, 7.1, 1.1$ Hz, 1H), 7.31 (d, $J = 8.3$ Hz, 1H), 7.27 (d, $J = 8.6$ Hz, 1H), 7.16 (ddd, $J = 7.9, 7.1, 1.1$ Hz, 1H), 7.01 (br s, 1H), 6.56 (t, $J = 5.8$ Hz, 1H), 4.16 (t, $J = 7.2$ Hz, 2H), 3.22–3.11 (m, 4H), 2.34–2.22 (m, 2H), 2.11–1.99 (m, 1H), 1.81–1.67 (m, 3H), 1.39–1.22 (m, 16H), 0.83 (t, $J = 6.9$ Hz, 3H). ^{13}C and DEPT NMR (101 MHz, CDCl_3) δ 169.83 (C=O), 168.57 (C=O), 142.34 (C), 141.14 (C), 126.45 (CH), 125.26 (C), 124.82 (CH), 122.94 (C), 122.73 (C), 120.68 (CH), 119.85 (CH), 119.68 (CH), 109.21 (CH), 108.50 (CH), 63.41 (CH₂), 60.41 (CH₂), 55.56 (CH₂), 50.89 (C), 47.36 (CH₂), 44.07 (CH₂), 36.67 (CH), 30.46 (CH₂), 29.53 (CH₂), 28.88 (CH₃), 20.58 (CH₂), 14.14 (CH₃). ESI: m/z 477.5 (M + H)⁺. HRMS calcd for $\text{C}_{29}\text{H}_{41}\text{N}_4\text{O}_2$ (M + H)⁺ 477.3229, found 477.3251.

N-[[1-(3,3-Dimethylbutyl)piperidin-4-yl]methyl]-9-pentyl-9H-carbazole-3-carboxamide (20). 9-Pentyl-9H-carbazole-3-carboxylic acid¹⁶ (100 mg, 0.27 mmol) was combined with cyanoborohydride (83.3 mg, 1.35 mmol/g) and 3,3-dimethylbutyraldehyde (0.06 mL, 0.31 mmol) and then suspended in anhydrous CH_2Cl_2 (2 mL). The mixture was stirred at room temperature overnight. The reaction mixture was diluted with DCM (30 mL), and the organic layer was washed three times with brine (50 mL). The organic layer was then separated, dried over magnesium sulfate, filtered, and concentrated in vacuo. The crude product was purified by flash chromatography eluting with heptane/EtOAc (0–100%) to yield the title compound (88.4 mg, 71%). ^1H NMR (400 MHz, CDCl_3) δ 8.51 (d, $J = 1.4$ Hz, 1H), 8.04 (d, $J = 7.8$ Hz, 1H), 7.86 (dd, $J = 8.6, 1.4$ Hz, 1H), 7.42 (ddd, $J = 8.2, 7.1, 1.1$ Hz, 1H), 7.35 (d, $J = 8.2$ Hz, 1H), 7.29 (d, $J = 8.6$ Hz, 1H), 7.15 (ddd, $J = 7.8, 7.1, 1.1$ Hz, 1H), 7.01 (br s, 1H), 6.59 (t, $J = 5.8$ Hz, 1H), 4.20 (t, $J = 7.2$ Hz, 2H), 3.36 (t, $J = 6.5$ Hz, 2H), 2.69–2.54 (m, 4H), 2.01 (td, $J = 11.7, 2.5$ Hz, 2H), 1.83–1.74 (m, 4H), 1.64–1.48 (m, 1H), 1.33–1.23 (m, 8H), 0.91 (s, 9H), 0.77 (t, $J = 6.9$ Hz, 3H). ^{13}C and DEPT NMR (101 MHz, CDCl_3) δ 168.76 (C=O), 142.47 (C), 141.23 (C), 126.67 (CH), 125.48 (C), 124.18 (CH), 122.76 (C), 122.53 (C), 120.68 (CH), 119.85 (CH), 119.68 (CH), 109.23 (CH), 108.50 (CH), 53.89 (CH₂), 52.31 (CH₂), 45.67 (CH₂), 43.42 (CH₂), 40.22 (CH₂), 36.21 (CH), 30.54 (CH₂), 30.12 (C), 29.42 (CH₂), 29.34 (CH₃), 28.71 (CH₂), 22.56 (CH₂), 14.03 (CH₃). ESI: m/z 462.3 (M + H)⁺. HRMS calcd for $\text{C}_{30}\text{H}_{43}\text{N}_3\text{O}$ (M + H)⁺ 462.3484, found 462.3452.

Physicochemical Properties. Molecular properties of compound **9** were calculated using Instant JChem 6.3.3, 2014, ChemAxon, Budapest, Hungary.

■ ASSOCIATED CONTENT

■ Supporting Information

Additional characterization data; ^1H and ^{13}C NMR spectra. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript. P.D. designed the compounds. P.D., S.W.M., and R.R.P. synthesized the compounds. C.B. carried out all electrophysiological experiments, and V.M.G. and N.D.B. carried out all animal testing. G.W.Z. and P.D. designed and supervised the study.

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Notes

The authors declare no competing financial interest.

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